

General Overview of Membranes & Applications

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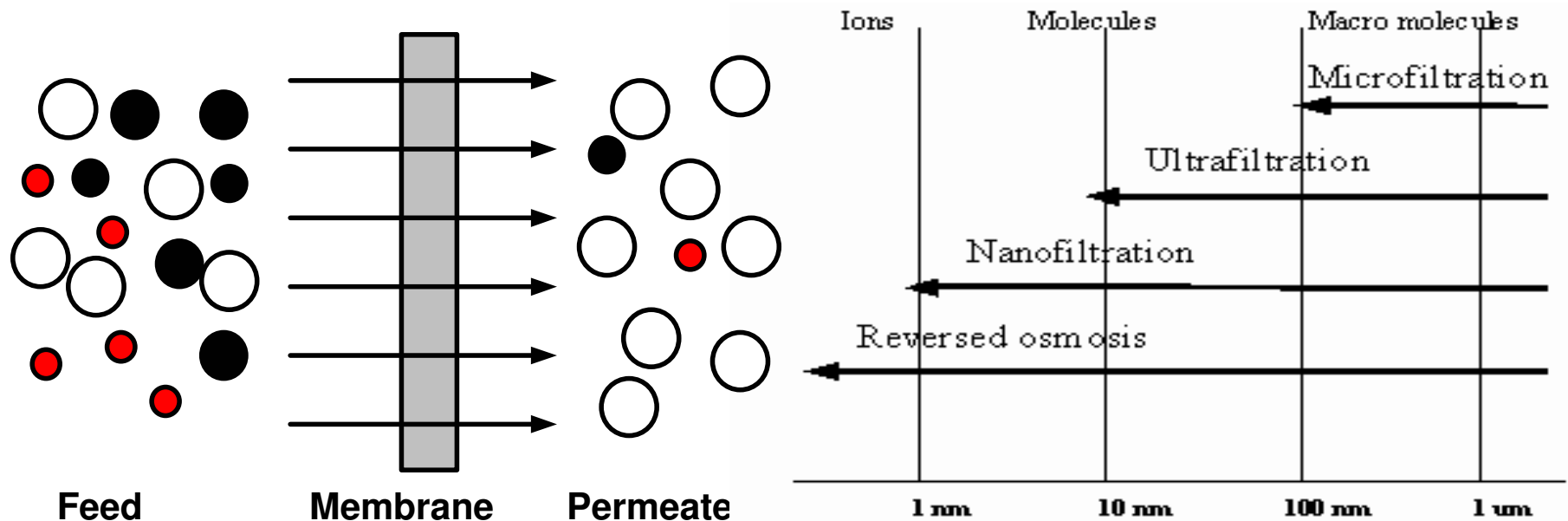
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Outline

- My Background
- Basic Membrane Definition
- Membrane Processes
- Applications
- Selection

Basic Membrane Definition

Semi-permeable materials that allow certain selected materials (solutes e.g. molecules or ions) to pass through while retaining others under a driving force.



What is a Membrane?

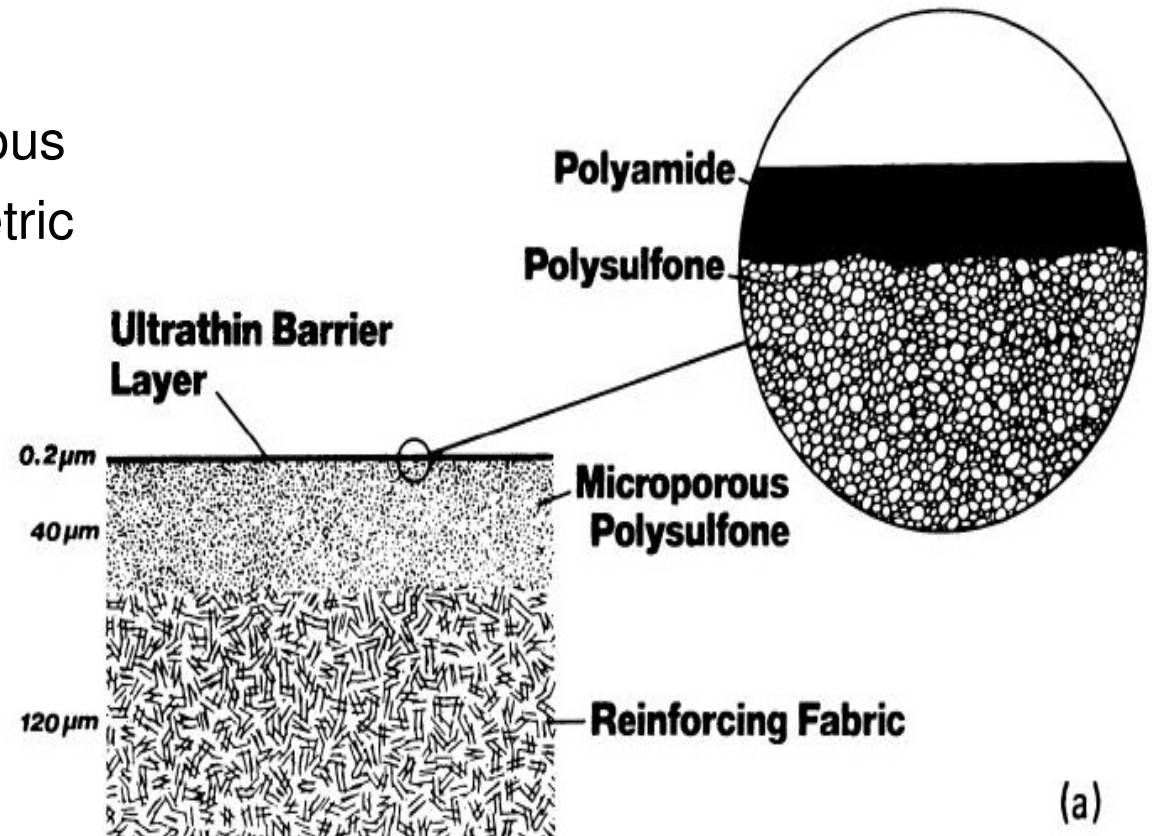
Intervening structure separating two phases and/or acting as an active or passive barrier to the transport of matter between the phases adjacent to it.

□ Structure:

- o Homo-.or heterogeneous
- o Symmetric or asymmetric
- o Solid or liquid

□ Charge

- o Negative
- o Positive
- o Neutral
- o Bipolar



(a)

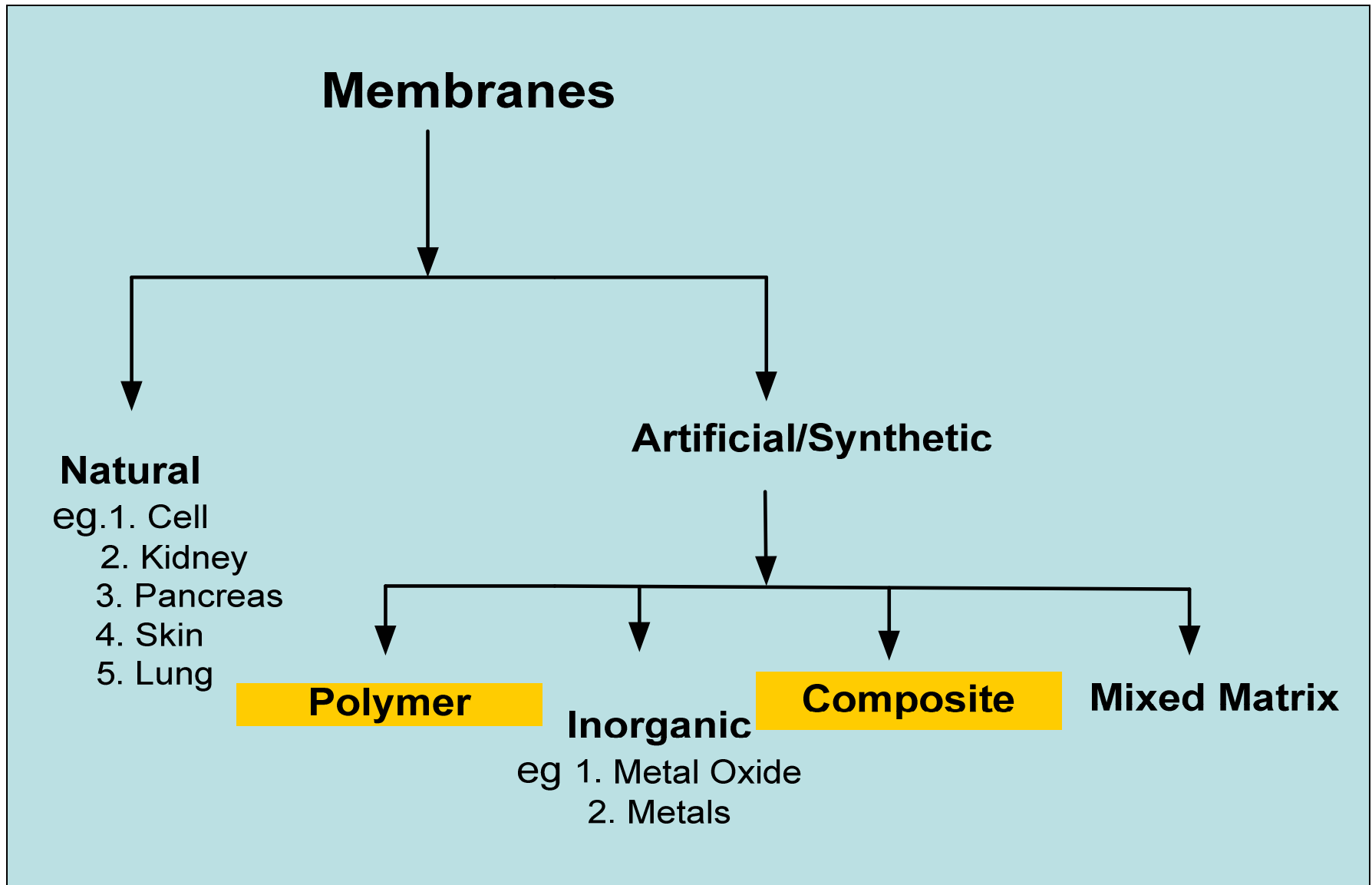
Some Advantages of Membranes

- Competing with niche technologies such as distillation, extraction, adsorption, stripping, degasification, etc
- Green process
- Lower energy consumption & foot print
- High selectivity e.g. for thermal labile products
- Wide range of applications
- Less sophisticated & ease of operation
- Modular expansion & ease of integration
- Linear scale from pilot and bench analysis

Some Disadvantages

- ❑ Newer Technology
- ❑ Fewer people with in-depth knowledge
- ❑ Could be expensive - polymer vs. ceramic
- ❑ Poor selection can heavily affect performance

Types of Membranes



Basic Functions of Membranes

- ❑ Separation e.g. water and ethanol purifications
- ❑ Discrimination e.g. pH meter, blood glucose meters
- ❑ Selective Transport e.g. contact lenses
- ❑ Delivery e.g. control drug delivery

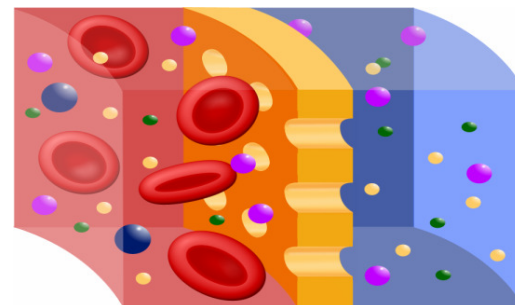
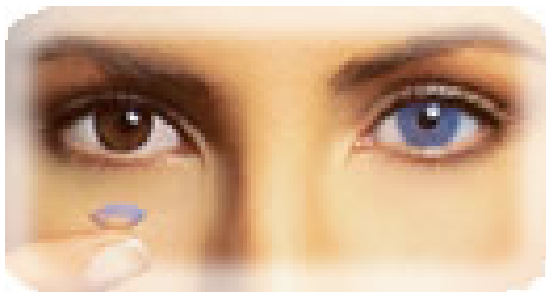
Membranes: Delivery Role

Control drug delivery e.g. patches such as nicotine patches, and some drugs, some oral or anal drugs, etc.



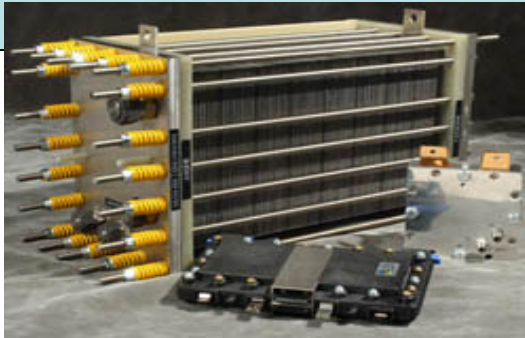
Membranes: Selective Transport Role

Applications includes contact lenses, artificial kidney, artificial liver and artificial pancreas, artificial lung (oxygenators), dialysis and hemodialysis.



Membranes: Discriminatory Role

Applications include batteries, fuel cells, and sensors. Sensors have been used for the detection of chemical and biological agents.



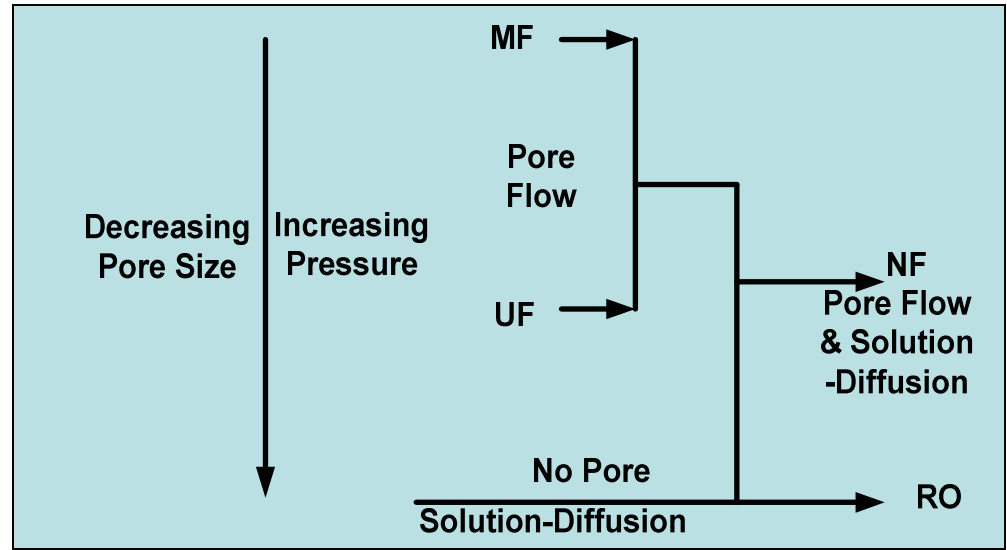
Membranes: Separation Role

- Water purification and waste water treatments
- Seawater desalination and brackish water purification
- Petroleum and petrochemical industries for separations
- Purification and recovery involving solvents.
- Sterilization: removal of viruses, bacteria, fungi and unwanted microbes



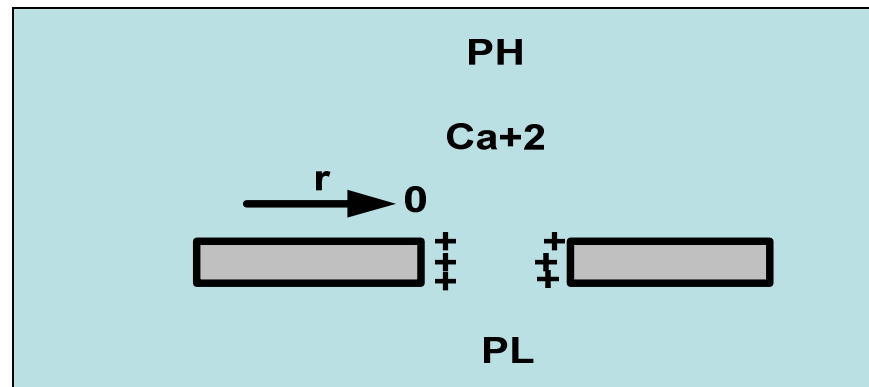
Some Common Membrane Processes

- Microfiltration (MF)
- Ultrafiltration (UF)
- Nanofiltration (NF)
- Reverse Osmosis (RO)
- Pervaporation (PV)
- Vapor/Gas Permeation
- Dialysis
- Hemodialysis
- Electrodialysis



$$E = k \frac{q_1 q_2}{r^2}$$

$$\Delta P = \frac{2\gamma \cos\beta}{r}$$

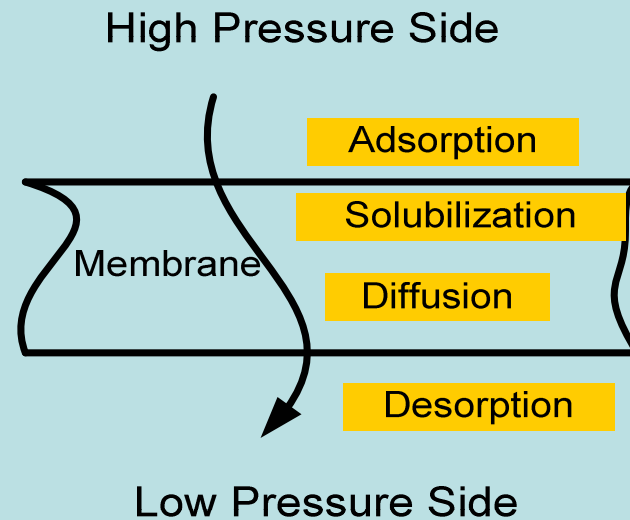


Separation Mechanisms

- ❑ Size
- ❑ Charge
- ❑ Shape
- ❑ Affinity
- ❑ Combination

Affinity- Cohesive Energy

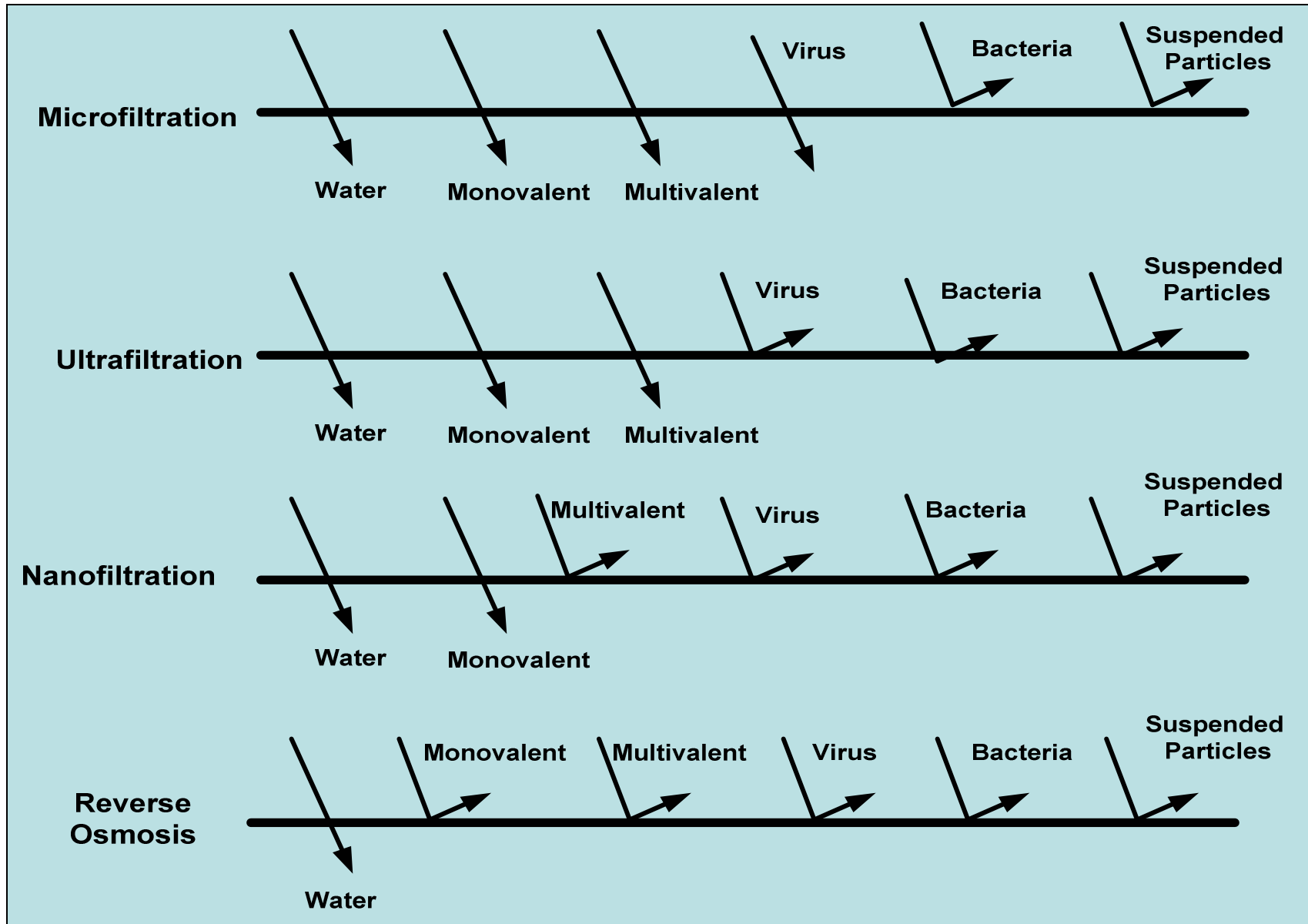
- o Adsorption
- o Solubilization
- o Diffusion
- o Desorption



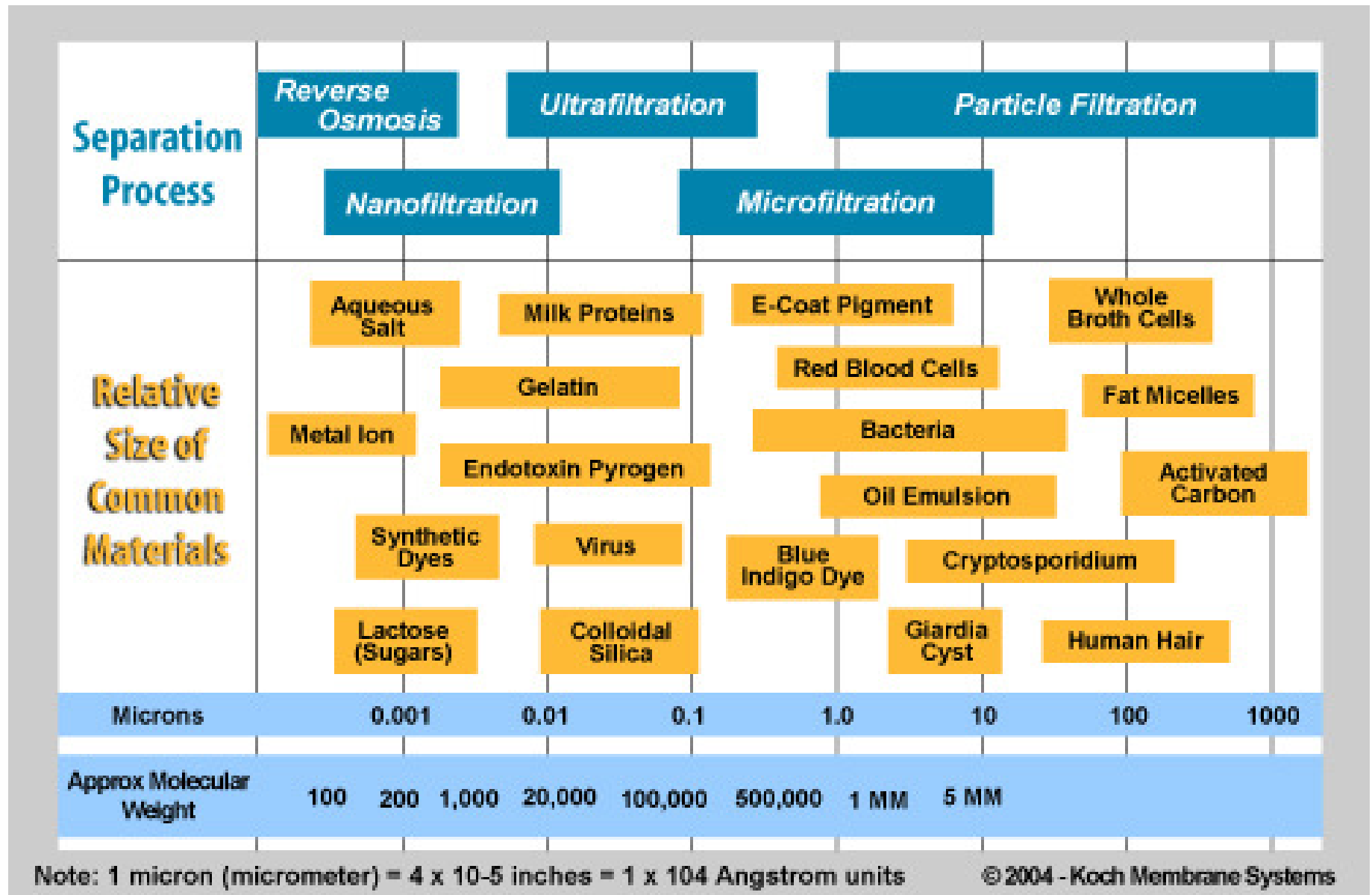
Driving Forces

- Pressure
- Concentration
- Partial Pressure
- Osmotic Pressure
- Electric Field
- Magnetic Field
- Temperature

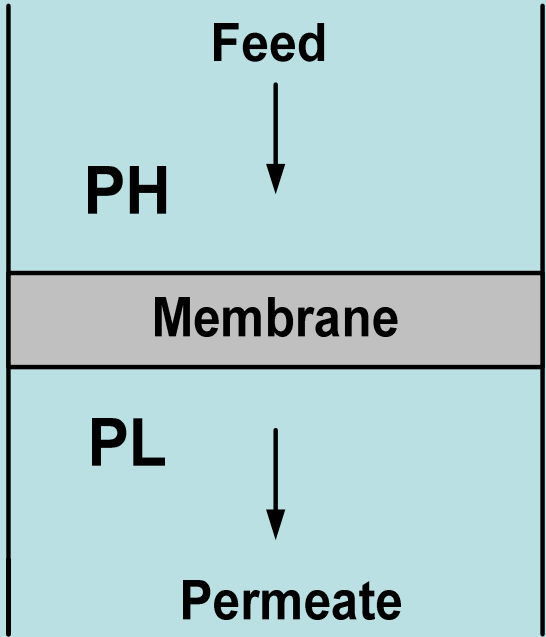
Membrane Process Capabilities



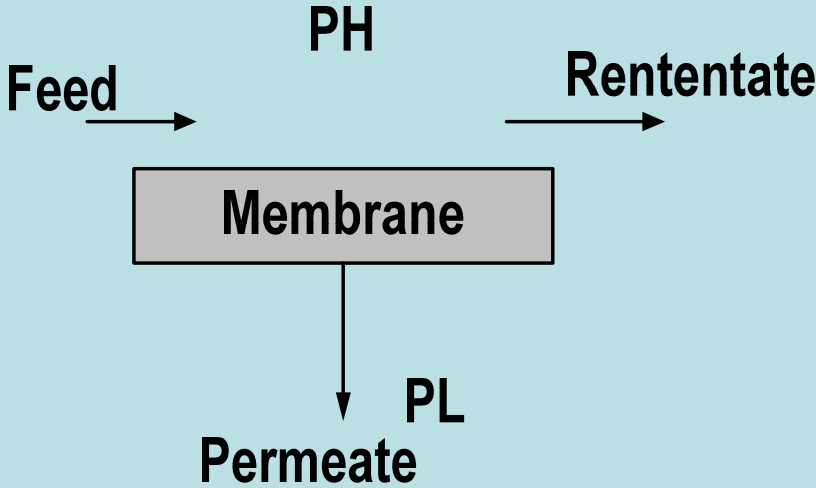
Membrane Process Capabilities



Flow Modes



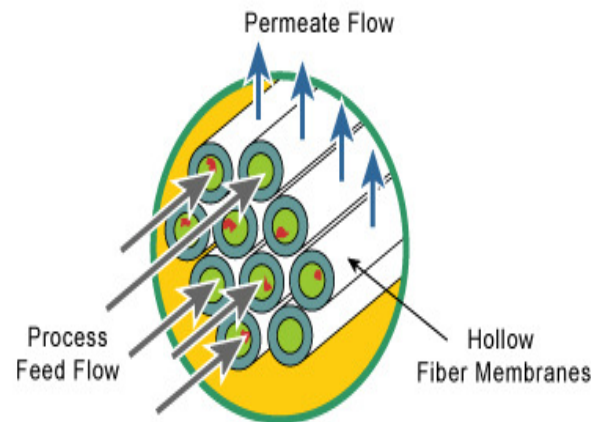
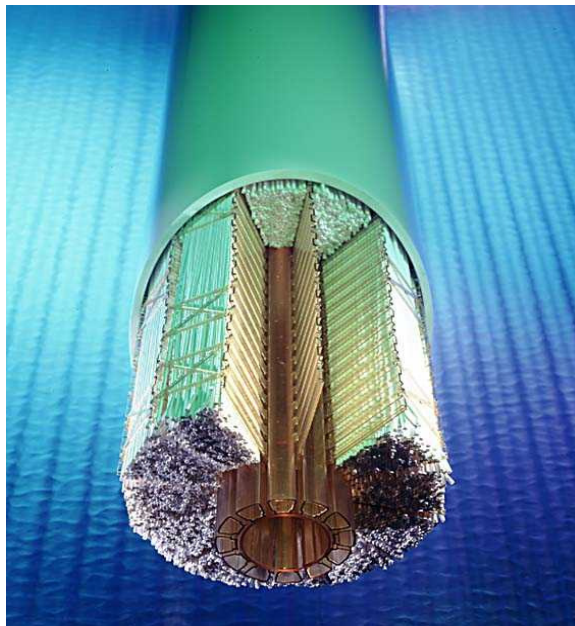
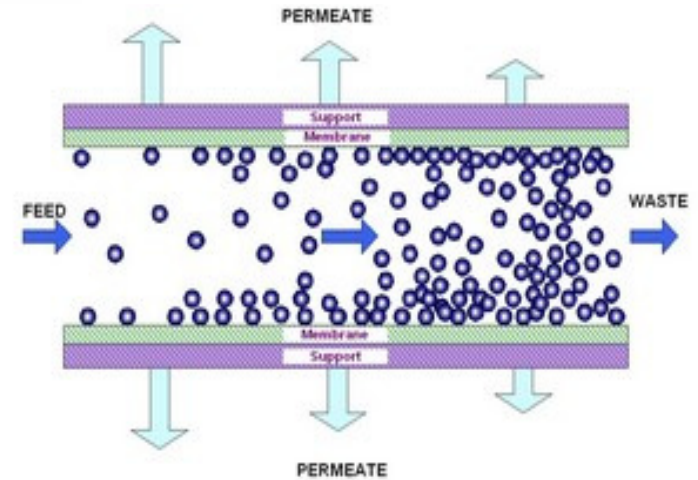
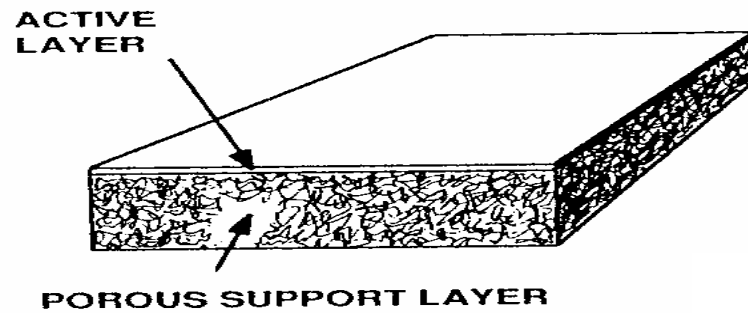
Dead-End



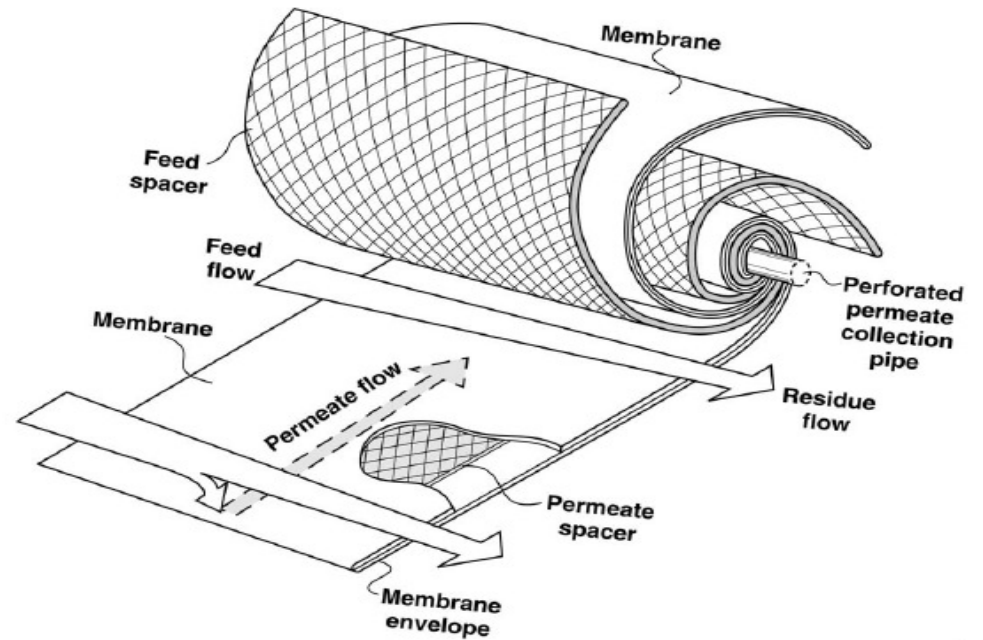
Cross Flow or Tangential

Configuration

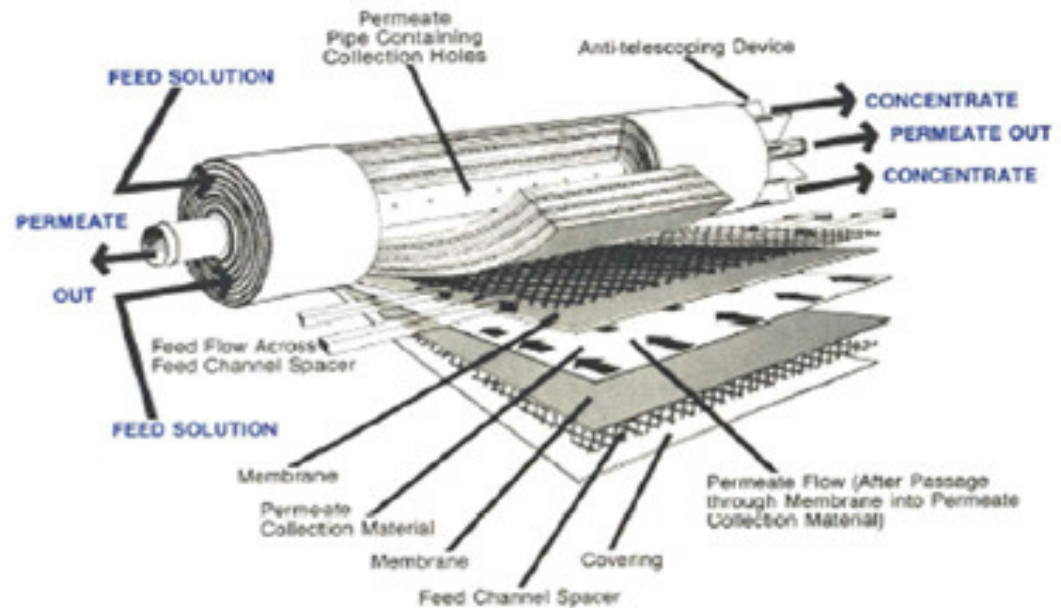
- Flat Sheet
- Tubular
- Hollow Fiber
- Spiral Wound



Configuration



Awesome Size. Awesome Savings.



□ Spiral Wound

Some Industrial Application of Membranes

Water and Pharmaceutical Purification

Waste Water Treatment

Seawater Desalination

Food & Beverage

- o Dairy processing
- o Clarification, Debittering
- o Bottled water
- o Dealcoholization of wine



Solvent Purification and Recovery

Biofuel Separation – Bioethanol, Biodiesel, etc

Petroleum & Petrochemical

- o Catalyst recovery, NPG purification
- o Petroleum & Petrochemical

Gas Separation & Purification

- o CO₂/H₂, NPG & LPG Purification

Membrane Applications: Petroleum & Petrochemical

- ❑ **Petrochemical**: Hydrocarbon and nitrogen recovery from vent streams in polypropylene production, ethylene oxide and vinyl acetate monomer production and polyvinyl chloride (PVC) production.



- ❑ **Natural Gas**: Fuel gas conditioning, natural gas liquids conditionings (NGL), recovery/dew point control, and nitrogen, carbon dioxide, and hydrogen sulfide removal



- ❑ **Refinery and Syngas**: Hydrogen purification and liquid petroleum gas (LPG) recovery from refineries, hydrogen separation in syngas processes and CO₂ removal from syngas.

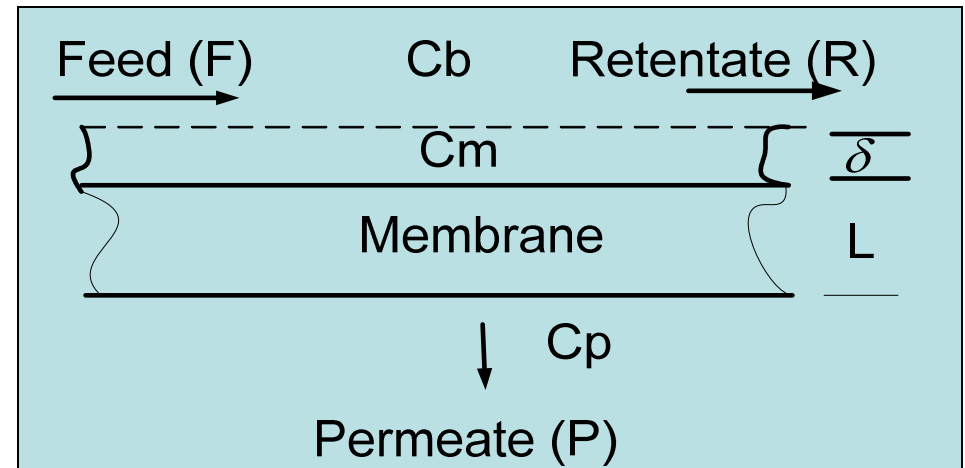


Propylene and Isobutane Recovery Membrane Unit at a Polyolefins Plant



Membrane Performance Characterization

$$1. \text{ Flux (q)} = \frac{\text{Volume}}{\text{Area} \times \text{Time}}$$



$$2. \text{ Permeability (Lp)} = \frac{\text{Flux (q)}}{\text{Transmembrane Pressure } (\Delta P)}$$

$$3. \text{ Observed Rejection (OR)} = \left[1 - \frac{C_p}{C_b} \right] \times 100 \%$$

$$4. \text{ Intrinsic Rejection (r)} = \left[1 - \frac{C_p}{C_m} \right] \times 100 \%$$

Factors Affecting Membrane Performance

Nature of the feed

- o Gas-liquid mixture
- o Liquid mixture
- o Solid and liquid mixture
- o Fouling feed, viscosity, density, etc

Flow Rate

Concentration

Temperature

Driving Force e.g. pressure, concentration, etc

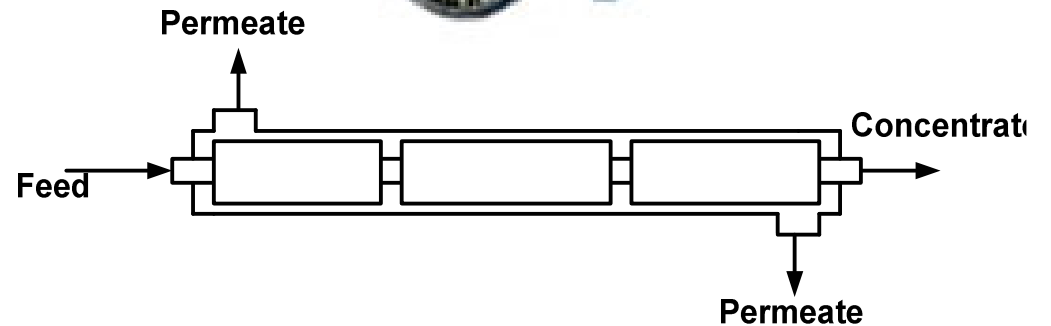
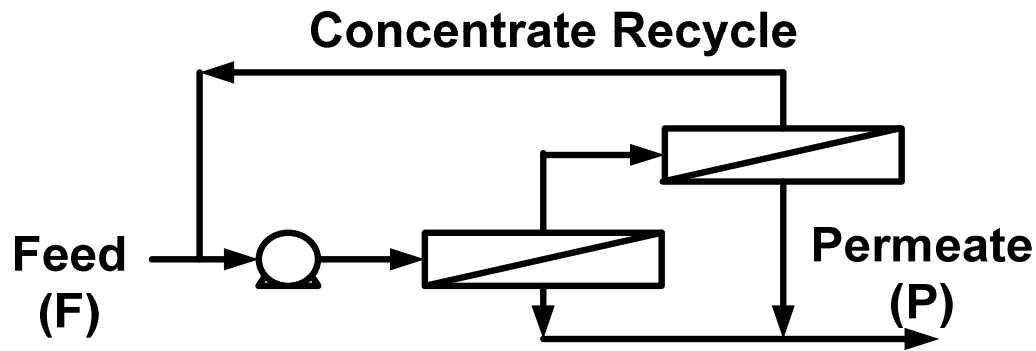
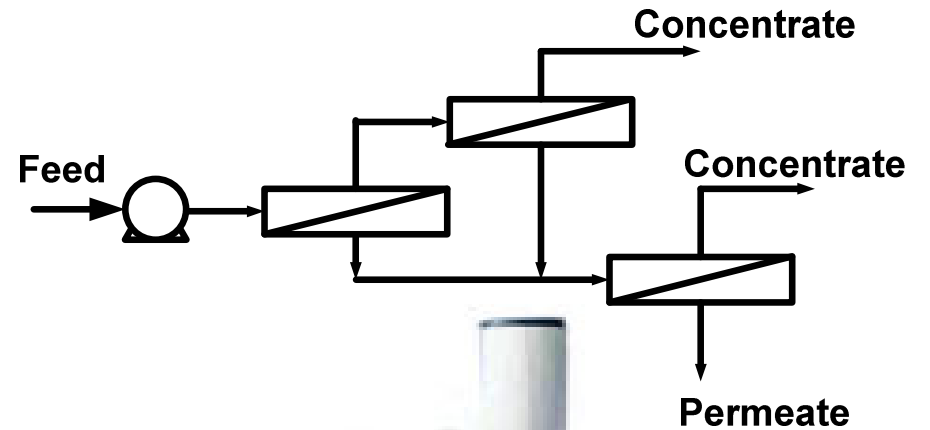
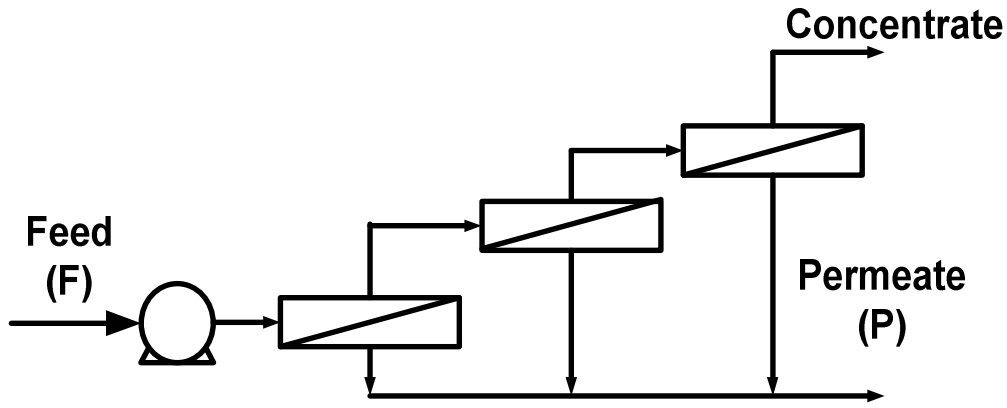
Type of Membrane Process and Material

Module Configuration

Problems Facing Membrane Performance

- ❑ Concentration Polarization
- ❑ Fouling
- ❑ Swelling
- ❑ Compaction

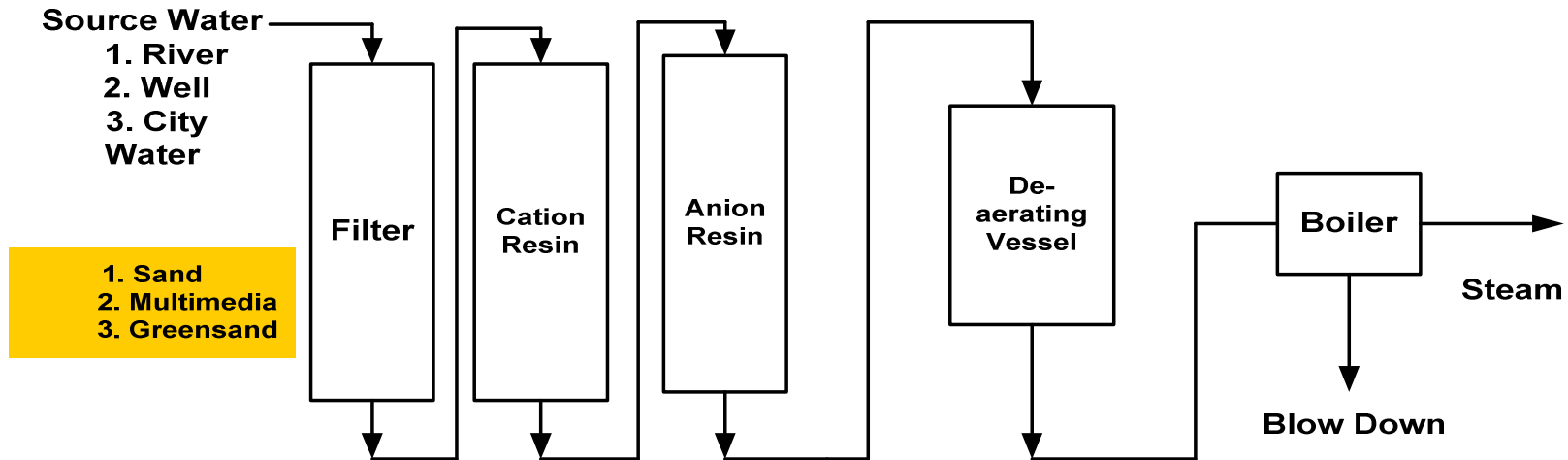
Membrane System Design



System Design Software Demo

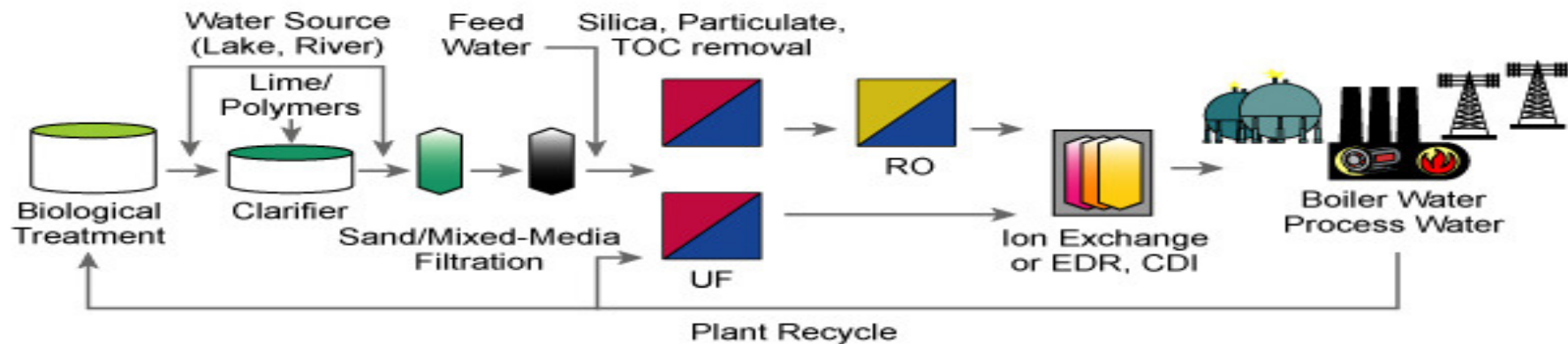
- For MF, UF, NF and RO
- Recovery (P/F)
- Stages
- Passes

Boiler Water Treatment Process



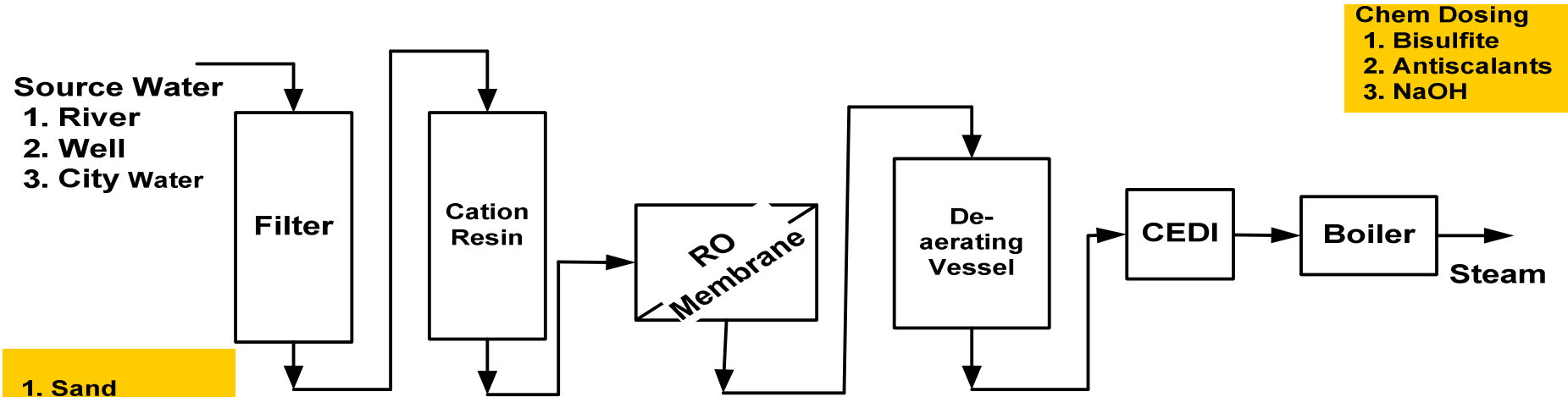
Cold lime softening before the filters depending on the hardness

Boiler Make-up Water for Power Plant



- Alternative to chemical and physical based RO pretreatment
- Provides higher quality product water for improved RO operation
- Option to reuse municipal wastewater or boiler blow down water

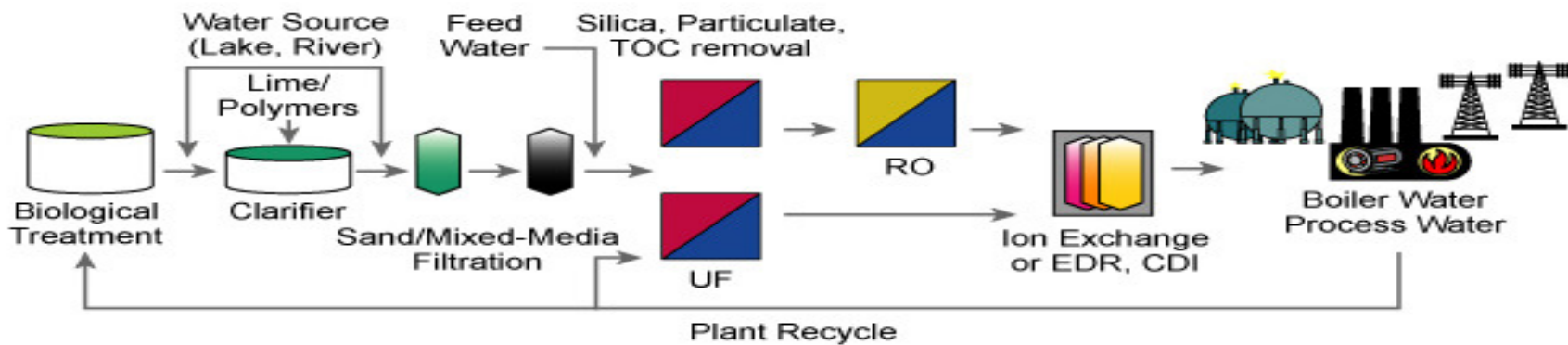
Boiler Water Treatment Process



Easier to treat well water river water

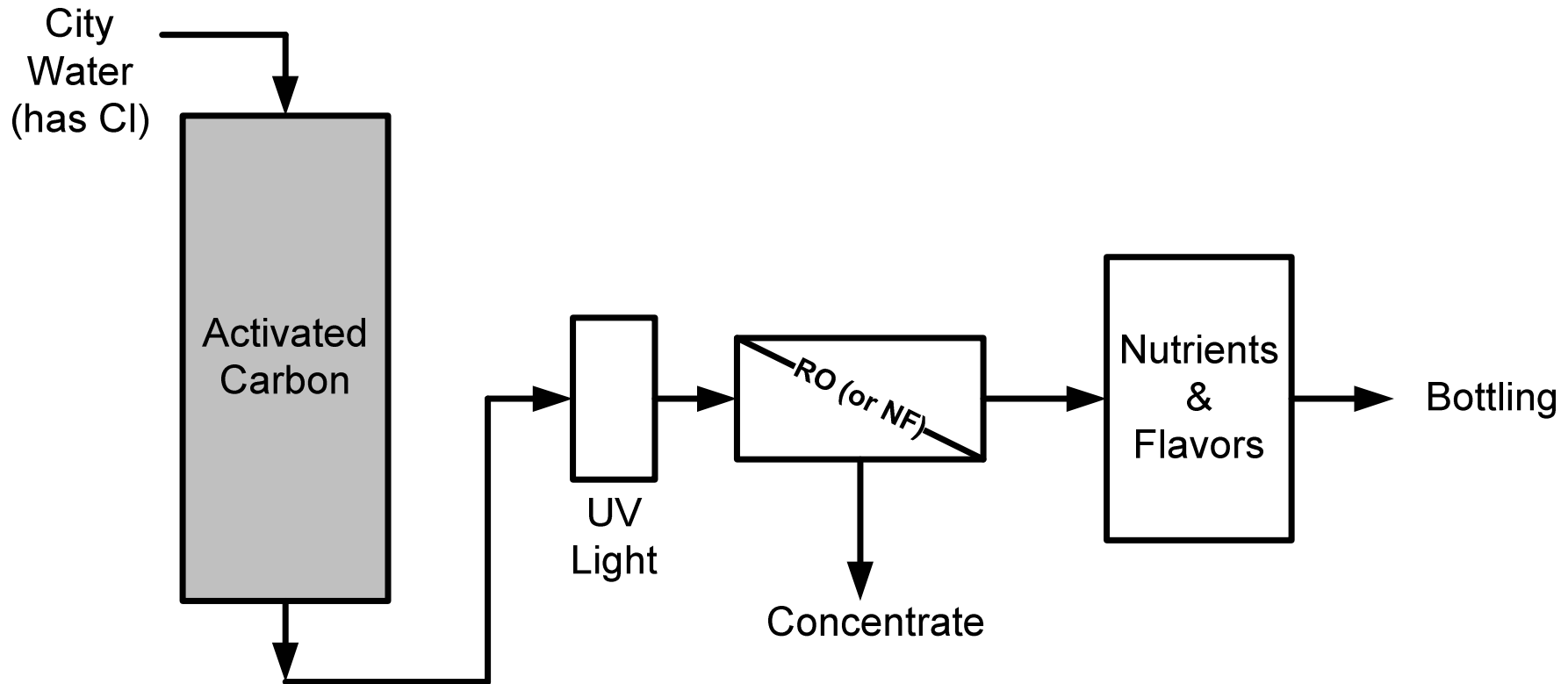
o First pass normally fed to demin. system
o Second pass fed to the boiler

Boiler Make-up Water for Power Plant

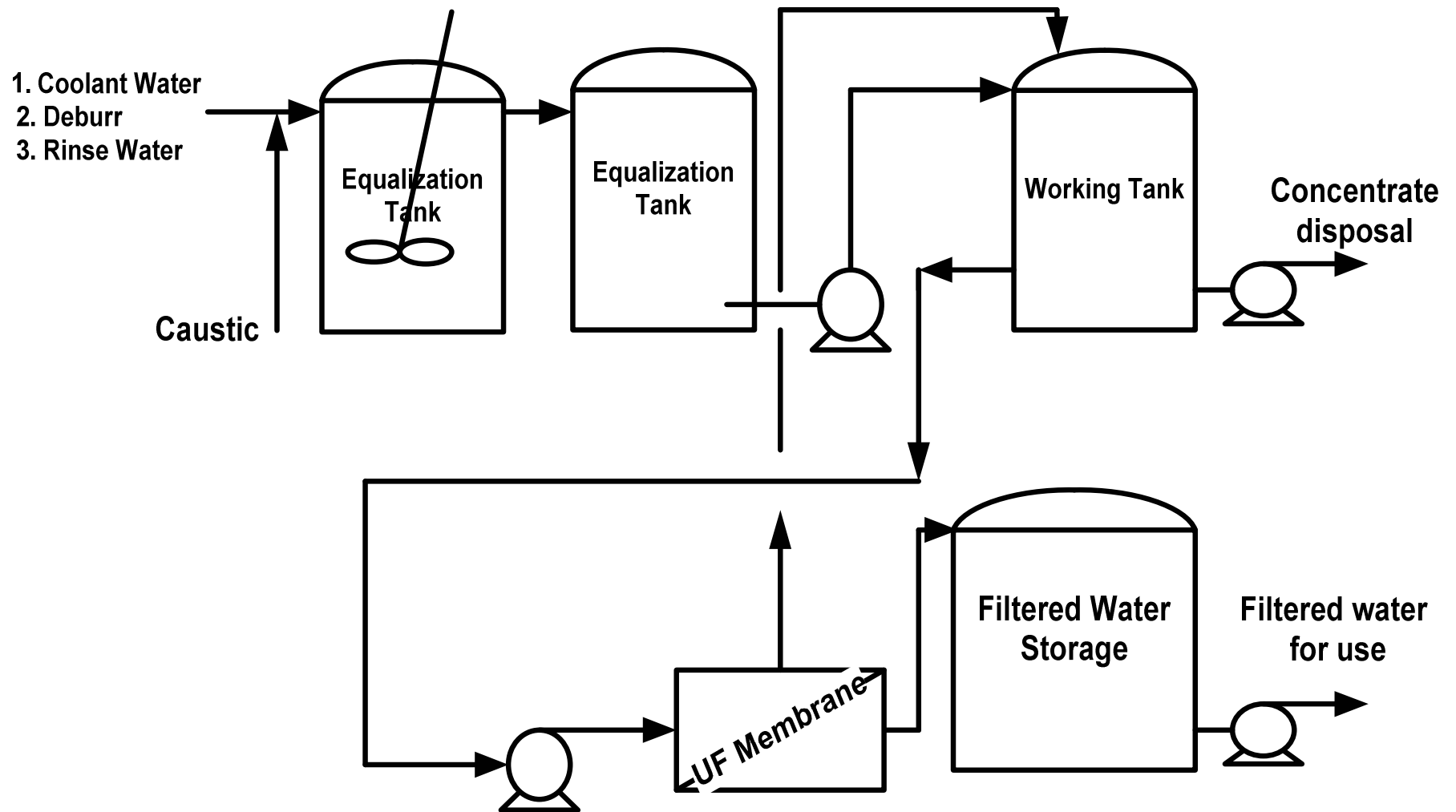


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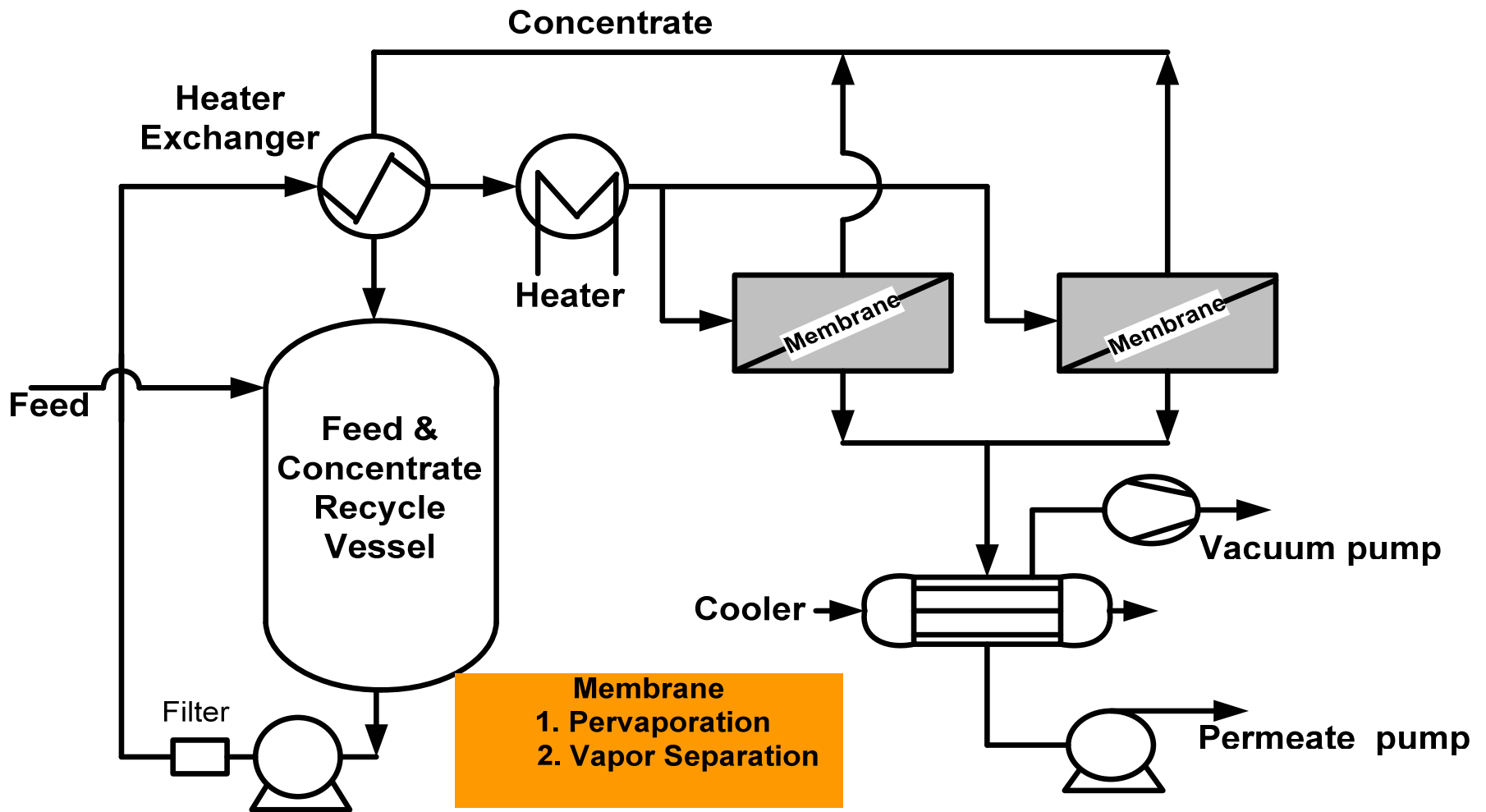
Purified Water Treatment Process



Waste Water Treatment for Re-use



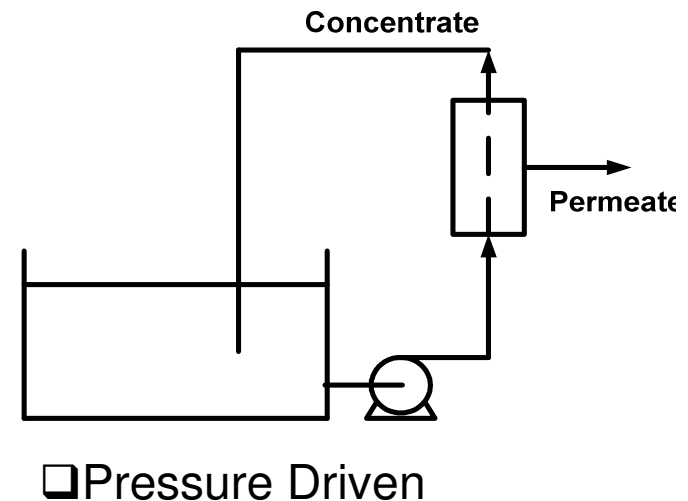
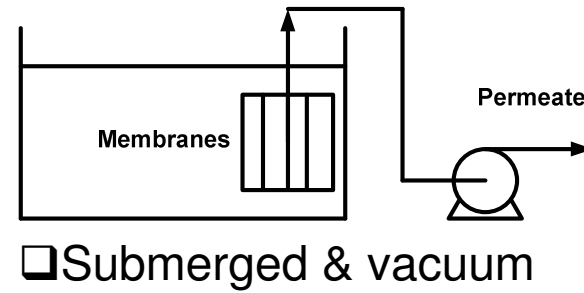
Biofuel & Solvent Dehydration & Purification



Desalination/Brackish Water Purification Process

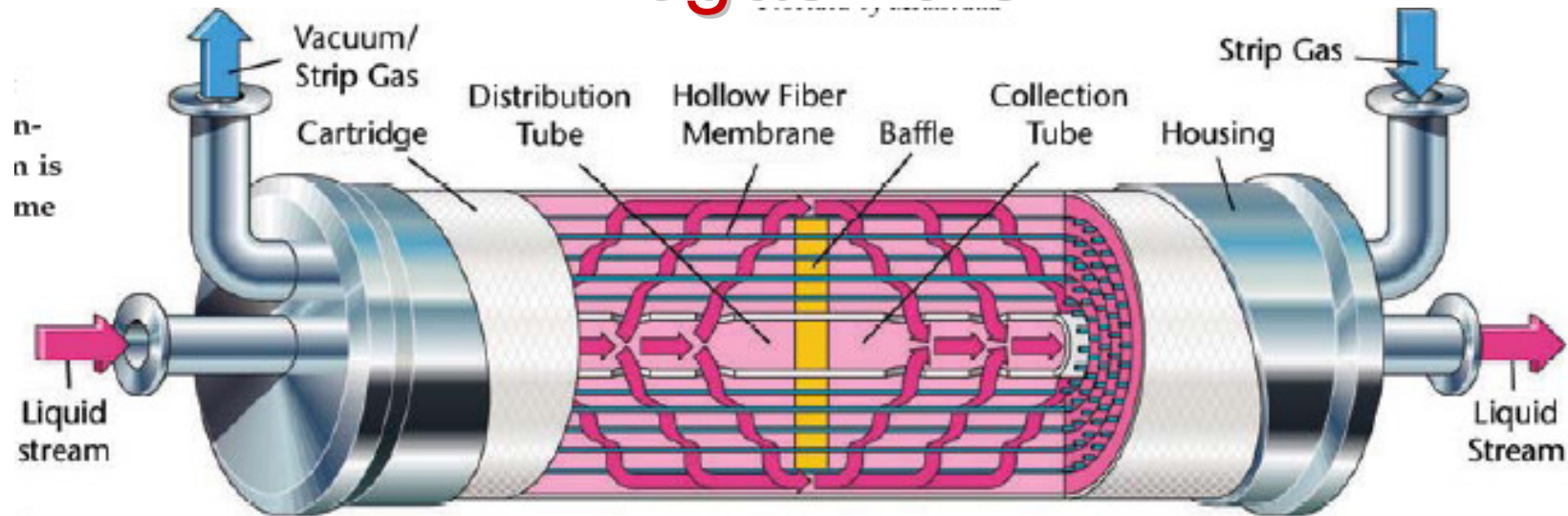


Membrane Bioreactors

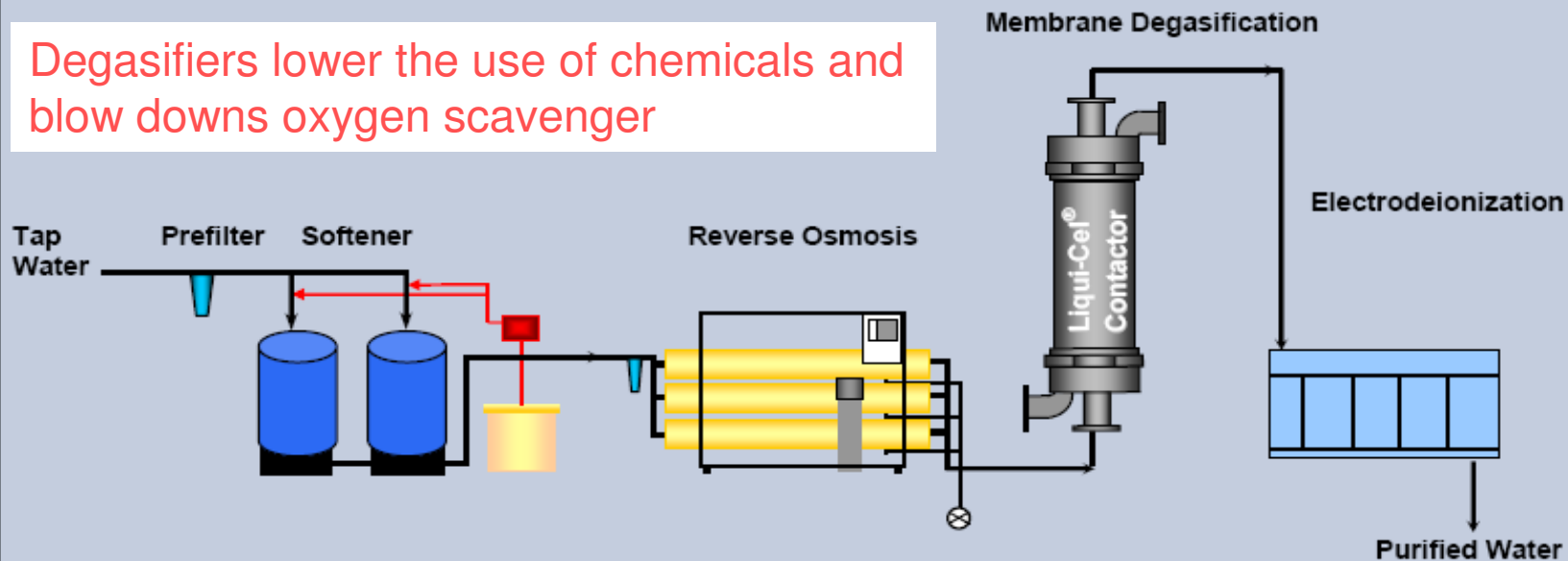


- For waste treatment e.g. organic waste, enzymatic or bacteria decomposition
- For biofuel separation e.g. Bioethanol & bio-butanol from fermentation broth
- For pharmaceutical separation application e.g. antibiotics from fermentation broth

Membrane Contactors: Gasification & Degasifiers

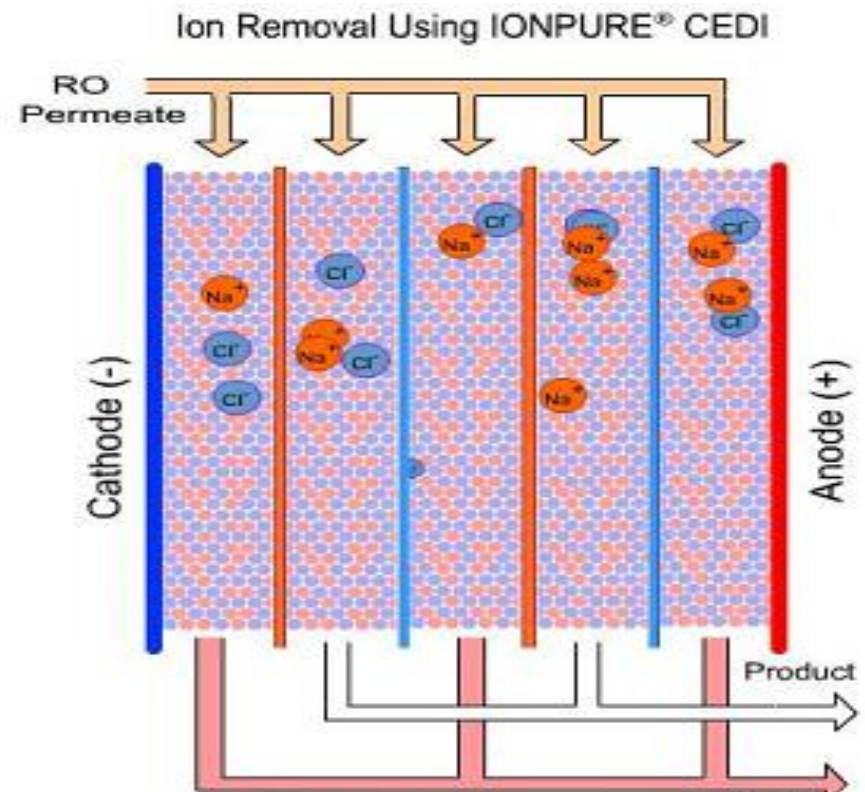
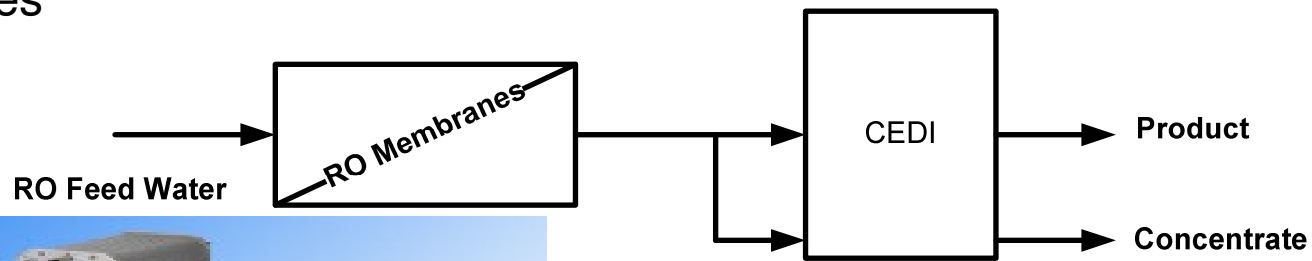


Degasifiers lower the use of chemicals and blow downs oxygen scavenger



Pharmaceutical and Microelectronic Grade Water

- ❑ RO membranes followed by Continuous Electrodeionization (CEDI)
- ❑ CEDI principles - Ion exchange resin + charged (cationic and anionic) membranes



Conclusions

- ❑ Membranes have wide and varied applications.
- ❑ Proper selection of membrane processes, materials, feed rate, driving forces and configuration are pertinent for designing an effective membrane system.
- ❑ Membranes also has very specialized applications.
- ❑ Could provide effective solution to the current quality water problems in Africa.



THANK YOU!

A cartoon illustration of a man with green hair, wearing a pink suit jacket, a white shirt, a pink tie, and blue trousers. He is waving with his right hand. To his right, the words "THANK YOU!" are written in large, bold, red, stylized letters with black outlines. The text is slanted upwards from left to right.

Questions ?



Large RO Systems



Degasifiers: Economic Benefits

Boiler Operating Details		
	Chemically Treated Feed Water	Degassed Feed Water
Boiler capacity	10,000 lb/hr	10,000 lb/hr
Pressure	50 psig	50 psig
Efficiency	80%	80%
Fuel	Natural Gas	Natural Gas
Fuel cost	4.5 USD/ 1000 ft ³	4.5 USD/ 1000 ft ³
Fuel efficiency	1000 BTU/ft ³	1000 BTU/ft ³
Condensate return	30%	30%
Boiler blow down rate	10%	5%
Hours of operation	6600 hrs/yr (275 days/yr)	6600 hrs/yr (275 days/yr)
Feed water costs	1.2 USD/1000 gallons	1.2 USD/1000 gallons
Sodium Sulfite cost	0.5 USD/lb	0.5 USD/lb
Feed water temperature	60° F	60° F
Inlet Dissolved	9.0 ppm	0.5 ppm

Comparison of Chemical Treatment System to Degassing System			
	Chemically Treated Feed Water	Degassed Feed Water	Savings
Chemical cost (US\$)	\$2,299.00	\$128.00	\$2,171.00
Blow down water costs	\$1,055.00	\$500.00	\$555.00
Energy cost due to heat loss in blow down	\$11,095.00	\$5,256.00	\$5,839.00
Total yearly costs/savings	\$13,997.00	\$5,669.00	\$8,565.00

Membrane Performance Characterization

$$q = \frac{Q}{A} = k \frac{\Delta P}{\mu L}$$

$$q = L_p(\Delta p - \sigma \Delta \pi)$$

$$\Delta \pi = \phi(C_m - C_p)RT$$

$$q = \frac{Q}{A}$$

$$\text{Permeability (} L_p \text{)} = \frac{\text{Flux (} q \text{)}}{\text{Transmembrane Pressure (} \Delta P \text{)}}$$

